Russian Journal of Organic Chemistry, Vol. 38, No. 6, 2002, pp. 902–904. Translated from Zhurnal Organicheskoi Khimii, Vol. 38, No. 6, 2002, pp. 944–945.

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## SHORT COMMUNICATIONS

## Iodosobenzene Dichloride, a Convenient Reagent for Chloroand Iodomethoxylation of Double and Triple Bonds<sup>\*</sup>

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Received January 16, 2002

Iodosobenzene dichloride (I), an organic compound of polyvalent iodine, is extensively used in organic synthesis for chlorination and oxidation of a wide range of compounds [1-3]. However published reports on application of the compound to haloalkoxylation are lacking.

We established that the reaction of  $PhICl_2$  with styrene derivatives **II–IV** in methanol at room temperature easily gave rise to products of electrophilic chloromethoxylation **V–VII** in 68–90% yield.

The chloromethoxylation proceeded as a typical electrophilic addition, and therewith, as was shown by an example of *trans*-stilbene (**III**), occurred predominantly *anti*-addition to furnish *erythro*-diastereomer **VI** in 64% yield (the yield of *threo*-isomer **VI** was 36%).

The reaction with  $\alpha$ -methylstyrene (**VIII**) was less selective and afforded in 70% overall yield an inseparable mixture of two isomers **IX** and **X** in 75:25 ratio according to GC-MS and <sup>1</sup>H NMR data. The formation of reaction product **X** is an evidence of a contribution from free-radical chlorination of the methyl group in compound **IX** [or in initial  $\alpha$ -methyl-styrene (**VIII**)] effected by PhICl<sub>2</sub>.

It turned out that reaction of styrene (II) with  $PhICl_2$  in methanol in the presence of iodine took another route easily providing a product of iodomethoxylation of the double bond XI in 80% yield. Obviously here the iodosobenzene dichloride acted as oxidant for iodine providing its electrophilic intermediates (I<sup>+</sup>, ICl, or CH<sub>3</sub>OI). This reaction is perhaps the first example of a simple iodoalkoxylation of alkenes with the use of compounds of polyvalent iodine, and it opens new opportunities for the synthesis of commonly less available iodoalkoxyalkanes [4, 5].

Phenylacetylene (**XII**) reacted with two moles of compound **I** in methanol furnishing 1,1-dimethoxy-1phenyl-2,2-dichloroethane (**XIII**) in 65% yield. Compound **XIII** was formerly prepared by reaction of



The study was carried out under financial support of the Russian Foundation for Basic Research (grant no. 00-03-32812a).

phenylacetylene with chlorine or *N*-chlorosuccinimide in methanol (yield 70–81%) [6].

Thus iodosobenzene dichloride provides a possibility to perform both chloro- and iodomethoxylation of aromatic alkenes and alkynes under mild conditions.

General procedure of the synthesis. To a solution of 2.0 mmol of alkene in 8.0 ml of MeOH was added 2.02 mmol of iodosobenzene dichloride (I), and the mixture was stirred at room temperature for 20 min. The mixture was poured into 30 ml of water, extracted with ether  $(2 \times 30 \text{ ml})$ , the extract was washed with water  $(2 \times 50 \text{ ml})$ , with saturated NaCl solution (50 m), and dried on Na<sub>2</sub>SO<sub>4</sub>. On removing the ether the residue was dissolved in benzene (2 ml) and applied to a column packed with silica gel. Elution with hexane afforded 390 mg of iodobenzene (95%) (bp 188°C [7]). The reaction products **V**, **VII**, **XIII** were eluted with hexane-benzene, 4:1.

**1-Methoxy-1-phenyl-2-chloroethane** (V). Yield 68%, oily substance, bp 82–85°C/hexane-benzene mixture, 4:1. 6 mm Hg (publ. bp 79–82°C/4 mm Hg [8]). <sup>1</sup>HNMR spectrum (500 MHz,  $CCl_4$ -CDCl<sub>3</sub>, 3:1, δ, ppm, *J*, Hz): 3.27 s (3H, OCH<sub>3</sub>), 3.50 d.d (1H, CH<sub>a</sub>, *J* 4.5, 11.5), 3.60 d.d (1H, CH<sub>b</sub>, *J* 7.5, 11.5), 4.27 d.d (1H, *J* 4.5, 7.5), 7.26–7.34 m (5H arom). <sup>13</sup>C NMR spectrum (500 MHz,  $CCl_4$ -CDCl<sub>3</sub>, 3:1, δ, ppm): 47.80 (CH<sub>2</sub>), 56.99 (OCH<sub>3</sub>), 83.79 (CH), 126.76, 128.35, 128.50, 138.72 (C arom). Found, %: C 63.40; H 6.43; Cl 20.04. C<sub>11</sub>H<sub>11</sub>ClO. Calculated, %: C 63.35; H 6.50; Cl 20.78.

1-Methoxy-1,2-diphenyl-2-chloroethane **(VI)**. Yield 90%. erythro-(VI) mp 98-100°C (publ. mp 95-97°C [8]). <sup>1</sup>H NMR spectrum (500 MHz, CDCl<sub>3</sub>, δ, ppm, J, Hz): 3.18 s (3H, OCH<sub>2</sub>), 4.53 d (1H, J7.0), 4.97 d (1H, J7.0), 7.24–7.29 m (10H arom). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, δ, ppm): 57.35 (OCH<sub>3</sub>), 65.32 (CHCl), 87.14 (CHOCH<sub>3</sub>), 127.89, 127.95, 128.24, 128.14, 137.71, 138.22 (C arom). Found, %: C 73.05; H 6.24; Cl 13.97. C<sub>15</sub>H<sub>15</sub>ClO. Calculated, %: C 73.02; H 6.13; Cl 14.37. threo-(VI) oily substance (publ. oily substance [8]). <sup>1</sup>H NMR spectrum (500 MHz, CDCl<sub>3</sub>, δ, ppm, J, Hz): 3.31 s (3H, OCH<sub>3</sub>), 4.46 d (1H, J 8.0), 4.95 d (1H, J 8.0), 7.17-7.22 m (10H arom). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>,  $\delta$ , ppm): 57.21 (OCH<sub>3</sub>), 66.75 (CHCl), 87.66 (CHOCH<sub>3</sub>), 127.58, 127.94, 127.99, 128.21, 137.47, 138.15 (C arom). Found, %: C 73.27; H 6.37; Cl 13.90. C<sub>15</sub>H<sub>15</sub>ClO. Calculated, %: C 73.02; H 6.13; Cl 14.37.

**1-Methoxy-1,1-diphenyl-2-chloroethane (VII).** Yield 72%, mp 47–49°C (publ. oily substance [8]). <sup>1</sup>H NMR spectrum (CCl<sub>4</sub>\_CDCl<sub>3</sub>, 3:1,  $\delta$ , ppm): 3.12 s (3H, OCH<sub>3</sub>), 4.22 s (2H), 7.19 t (4H arom) 7.22 t (6H arom), 7.30 (4H arom). <sup>13</sup>C NMR spectrum (CCl<sub>4</sub>\_CDCl<sub>3</sub>, 3:1,  $\delta$ , ppm): 48.36 (CH<sub>2</sub>Cl), 50.57 (OCH<sub>3</sub>), 81.69 (COCH<sub>3</sub>), 127.14, 127.30, 128.05, 142.43 (C arom). Found, %: C 72.96; H 6.47; Cl 13.70. C<sub>15</sub>H<sub>15</sub>ClO. Calculated, %: C 73.02; H 6.13; Cl 14.37.

**1,1-Dimethoxy-1-phenyl-2,2-dichloroethane** (**XIII**). Yield 65%, mp 68–69°C (publ. mp 66– 67.5°C [6]). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>,  $\delta$ , ppm): 3.34 s (6H, OCH<sub>3</sub>), 5.97 s (1H), 7.39 m (3H arom) 7.62 m (2H arom). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>,  $\delta$ , ppm): 49.89 (OCH<sub>3</sub>), 73.10 (CCl<sub>2</sub>), 102.81 [C(OCH<sub>3</sub>)<sub>2</sub>], 127.86, 128.89, 129.22, 133.84 (C arom). Found, %: C 50.87; H 5.05; Cl 29.87. C<sub>10</sub>H<sub>12</sub>Cl<sub>2</sub>O<sub>2</sub>. Calculated, %: C 51.09; H 5.14; Cl 30.16.

2-Iodo-1-methoxy-1-phenylethane (XI). To a solution of 267 mg (1.05 mmol) of iodine in 8 ml of MeOH was added 275 mg (1.0 mmol) of iodosobenzene dichloride, and the mixture was stirred for 5 min at room temperature. The mixture was poured into a solution of 208 mg (2.0 mmol) of styrene (II) in 2.0 ml of MeOH, and the mixture was stirred at room temperature for 10 min. Then it was poured into 30 ml of 5% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>. Further workup was as described above. We obtained 420 mg (80%) of oily compound (publ. oily compound [5]). <sup>1</sup>H NMR spectrum [400 MHz, CCl<sub>4</sub>(CD<sub>3</sub>)<sub>2</sub>CO, 3:1, δ, ppm, *J*, Hz]: 3.21 s (3H, OCH<sub>3</sub>), 3.23 m (2H, CH<sub>2</sub>), 4.21 d.d (1H, *J* 5.0, 8.0), 7.26–7.34 m (5H arom). <sup>13</sup>C NMR spectrum [CCl<sub>4</sub>(CD<sub>3</sub>)<sub>2</sub>CO, 3:1, δ, ppm]: 10.10 (CH<sub>2</sub>I), 56.94 (OCH<sub>3</sub>), 83.55 (CH), 126.51, 128.23, 128.53, 139.98 (C arom). Found: m/z 261.98537  $[M]^+$ .  $C_{9}H_{11}IO$ . Calculated: M 261.98564. Mass spectrum (Electron impact, 70 ev), m/z ( $I_{rel}$ , %): 262 (>1), 135 (14), 121 (100), 104 (10), 103 (9), 91 (10), 77 (13).

NMR spectra were registered on spectrometer Bruker AM-400 (400.13 MHz for <sup>1</sup>H, 100.61 MHz for <sup>13</sup>C spectra) for compound **XI**, the NMR spectra of all other compounds were registered on spectrometer Bruker DRX-500 (500.13 MHz for <sup>1</sup>H, 125.76 MHz for <sup>13</sup>C spectra). NMR and mass spectra were measured and elemental analyses were carried out at TsKP, Siberian Division, Russian Academy of Sciences (grant of the Russian Foundation for Basic Research no. 00-03-40135).

RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 38 No. 6 2002

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